

# Introduction To Phase Equilibria In Ceramic Systems By Hummel

By Hummel

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Aug 06, 2010 This is the introduction to the course "Phase and Chemical Equilibrium" that is taught at the graduate level in the Department of Chemical Engineering

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<http://www.sciencedirect.com/science/article/pii/S0925838896031210>

Introductory Chemical Engineering Thermodynamics, problem-solving strategies for energy balances and phase equilibria, 10.1 Introduction to Phase Diagrams 370.

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The equilibrium constant,  $K$ , expresses the relationship between products and reactants of a reaction at equilibrium with respect to a specific unit.

[http://chemwiki.ucdavis.edu/Physical\\_Chemistry/Equilibria/Chemical\\_Equilibria/The\\_Equilibrium\\_Constant](http://chemwiki.ucdavis.edu/Physical_Chemistry/Equilibria/Chemical_Equilibria/The_Equilibrium_Constant)

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